

## 46 Ultimate Chemical Equations Handbook Answers

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ROUND 2 Exercise 7-2: Predict and balance the following synthesis and decomposition reactions. Use abbreviations to 1. 2. 3. 4. 5. 6. 8. 9. 10.

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The Ultimate Chemical Equations Handbook Determine the oxidation number of the underlined element:  $\text{KMnO}_4$  Since K is an alkali metal, its charge must be 1+ Oxygen is 2- but there are four of them, therefore, 4 times 2- equals 8- If 1+ and 8- are added Ultimate Chemical Equation Answers

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