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1 1 1.1 Basic concepts: atoms The notation: $^{50}_{24}\text{Cr}$ shows that the atomic number, Z , is 24 and the mass number for the isotope is 50. Number of protons = Number of electrons = $Z = 24$ Number of neutrons = Mass number - $Z = 50 - 24 = 26$ For each isotope, $Z = 24$ and so there are 24 electrons and 24 protons.

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