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Chapter 1 Section 2 MATTER and Its  
Properties Matter – Anything that  
has MASS and VOLUME. ATOMS and  
MOLECULES the basic building blocks  
of matter. ATOM-- the smallest unit of  
an element that still maintains the  
properties that element. ELEMENT–  
a pure substance made up of only one  
type of atom.

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Section 1

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completes the statement or answers the question. 1. Atomic radii cannot be measured ...

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Explain why the flow of a compressed  
gas must be controlled for practical  
and safe use.

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Changes Matter—Properties and  
Changes  
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accepted 5. Dmitri Mendeleev 6.  
atomic mass 7. elements 8. atomic  
number 9. Henry Moseley 10. protons  
11. periodic law 12. properties 13. 14  
...

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Chapter 5 25 Electrons in Atoms  
Reviewing Vocabulary Match the  
definition in Column A with the term  
in Column B. Column A Column B 1.



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The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

## Chapter Assessment

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is, energy produced or absor... A

system that can exchange neither  
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Chapter 11 31 number of moles of

known substance  $C_3H_8CO_2$

$CO_2$  number of

moles of unknown substance  $CO_2$

$C_3H_8CO_2$  use

mole ratio 5 mol  $O_2$  1 mol  $C_3H_8$

3 mol  $CO_2$  1 mol  $C_3H_8$  4 mol  $H_2O$

1 mol  $C_3H_8$  1 mol  $C_3H_8$  5 mol  $O_2$

## MATH SKILLS TRANSPARENCY

### MASTER 16

A “ chemical change ” is a change that produces matter with a different composition than the original matter.

Chemical Change A change in which one or more substances are converted into different substances.

Heat and light are often evidence of a chemical change. Properties of

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Compounds have quite different properties than their component elements.

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