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Biology Chapter 3: Section 3-2 Review: Carbon Compounds ...

SECTION 3-2 REVIEW CARBON COMPOUNDS Date VOCABULARY REVIEW Define the following terms and provide one example for each. 1. organic compound 2. functional group 3. alcohol 4. monomer 5. polymer MULTIPLE CHOICE Write the correct letter in the blank. 1. Organic compounds contain a. carbon and usually other elements.

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Name Class SECTION 3-2 REVIEW CARBON COMPOUNDS Date VOCABULARY REVIEW Define the following terms and provide one example for each. 1. organic compound 2. functional group 3. alcohol 4. monomer 5. polymer MULTIPLE CHOICE Write the correct letter in the blank. 1. Organic compounds contain a. carbon and usually other elements.

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Section 3-2 VOCABULARY REVIEW 1. An organic compound is a compound containing carbon atoms covalently bonded to other carbon atoms and to other elements. Examples: any carbon-containing compound, such as benzene, ethanol, glycerol, glucose, fructose, sucrose, ATP, and ADP. 2.A

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The Section is built for long, fast rides, when you're in it for more than just the scenery. This All-Road rocket is built around big tires and a lightweight frame with geometry and features that won't let any road interrupt the rhythm of a great day out. Its lateral stiffness and shock-absorbing vertical compliance allow for efficient power transfer while smoothing out rough surfaces so ...

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SECTION 3-2 REVIEW Date VOCABULARY REVIEW Distinguish between the terms in each of the following pairs of terms. mbMo 51 mPLE SUCA-R 1. monosaccharide, polysaccharide CA-ZBOM ox 66/V PO c 2. amino acid, protein h-c-/ò Co o K M/ c L/ã-/ø--s 3. nucleotide, nucleic acid A- A- R IA./C MULTIPLE CHOICE Write the correct letter in the blank. 1.

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Section 2 3 Carbon Compounds (pages 44 48) This section explains how the element carbon is able to form millions of carbon, or organic, compounds. It also describes the four groups of organic compounds found in living things. The Chemistry of Carbon (page 44) 1. How many valence electrons does each carbon atom have? Each carbon atom has four electrons. 2.

Macromolecules The Chemistry of Carbon

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3. 13. This allows carbon atoms to form a tremendous variety of very large and complex molecules. Figure 3.13 Carbon bonding: these structures form the backbone of many different types of organic molecules. Organic Compounds Figure 3.12 This model represents a molecule of glucose, an organic compound composed of carbon, hydrogen, and oxygen.

Chapter 3: The Chemical Basis for Life Lesson 3.2: Organic ...

Section Review 3-2 1. The sun is the main source of energy used by life on Earth. 2. Some organisms use the energy stored inside inorganic compounds. 3. Energy flows through an ecosystem in one direction, from the sun to autotrophs and then to heterotrophs. 4. About 10% of the energy at any level is available to the next level. 5.

Ch. 3 Answer Key - Lawndale High School

c. only carbon. d. only carbon and hydrogen. 2. The number of covalent bonds a carbon atom can form with other atoms is 3. A covalent bond formed when two atoms share two pairs of electrons is called a a. -single bond. b. double bond. 4. The breakdown of a polymer involves a. hydrolysis.

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Section 203 203 Carbon Compounds

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Section 3-2 VOCABULARY REVIEW 1. An organic compound is a compound containing carbon atoms covalently bonded to other carbon atoms and to other elements. Examples: any carbon-containing compound, such as benzene, ethanol, glycerol, glucose, fructose, sucrose, ATP, and ADP.

SECTION 3-3 REVIEW MOLECULES OF LIFE

SECTION 3-1 REVIEW CARBON COMPOUNDS VOCABULARY REVIEW Define the following terms and provide one example for each. 1. organic compound 2. functional group 3. alcohol 4. monomer 5. polymer MULTIPLE CHOICE Write the correct letter in the blank. 1. Organic compounds contain a. carbon and usually other elements. b. many kinds of elements except ...

SECTION 3-1 REVIEW CARBON COMPOUNDS

Superfund 2019 Annual Accomplishments Report. EPA has released Superfund's annual report covering major accomplishments and environmental progress during fiscal year 2019.

Superfund | US EPA

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